Name	Date	Class

Elements and the Periodic Table • Skills Lab

S 8.7.a, 8.9

Alien Periodic Table

Problem

Imagine that inhabitants of another planet send data about 30 elements to Earth. However, these inhabitants use different names and symbols for these elements than humans do. Which elements on the periodic table do these "alien" names represent?

Skills Focus

drawing conclusions, classifying, interpreting data, inferring

Materials

ruler

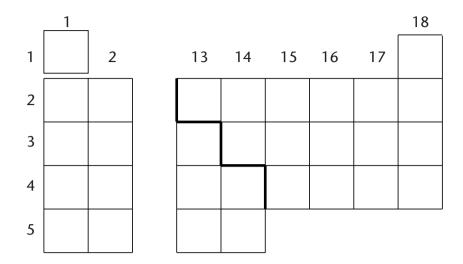
periodic table from text for reference

Procedure

- 1. Be sure to record your data in the blank periodic table on the next page.
- **2.** Listed below are data on the chemical and physical properties of the 30 elements. Place the elements in their proper position in the blank periodic table.
 - The inert gases are **bombal** (Bo), **wobble** (Wo), **jeptum** (J), and **logon** (L). Among these gases, wobble has the greatest atomic mass and bombal the least. Logon is lighter than jeptum.
 - The most reactive group of metals are **xtalt** (X), **byyou** (By), **chow** (Ch), and **quackzil** (Q). Of these metals, chow has the lowest atomic mass. Quackzil is in the same period as wobble.
 - **Apstrom** (A), **vulcania** (V), and **kratt** (Kt) are nonmetals whose atoms typically gain or share one electron. Vulcania is in the same period as quackzil and wobble.
 - The semimetals are ernst (E), highho (Hi), terriblum (T), and sississ (Ss). Sississ is the metalloid with the greatest atomic mass. Ernst is the semimetal with the lowest atomic mass. Highho and terriblum are in Group 14. Terriblum has more protons than highho. Yazzer (Yz) touches the zigzag line, but it's a metal, not a semimetal.
 - The lightest element of all is called **pfsst** (Pf). The heaviest element in the group of 30 elements is **eldorado** (El). The most chemically active nonmetal is apstrom. Kratt reacts with byyou to form table salt.

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Alien Periodic Table



- The element **doggone** (D) has only 4 protons in its atoms.
- Floxxit (Fx) is important in the chemistry of life. It forms compounds made of long chains of atoms. Rhaatrap (R) and doadeer (Do) are metals in the fourth period, but rhaatrap is less reactive than doadeer.
- Magnificon (M), goldy (G), and sississ are all members of Group 15. Goldy has fewer total electrons than magnificon.
- Urrp (Up), oz (Oz), and nuutye (Nu) all gain 2 electrons when they react. Nuutye is found as a diatomic molecule and has the same properties as a gas found in Earth's atmosphere. Oz has a lower atomic number than urrp.
- The element **anatom** (An) has atoms with a total of 49 electrons. **Zapper** (Z) and **pie** (Pi) lose two electrons when they react. Zapper is used to make lightweight alloys.

Analyze and Conclude

Write your answers on a separate sheet of paper.

1.	Drawing Conclusions List the Earth names for the 30 alien elements in
	order of atomic number.

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ΑI	ien Periodic Table	(continued)			
2.	Classifying Were you able to place some elements within the periodic table with just a single clue? Explain using examples.				
3.	Interpreting Data Why did you		e clues to place other		
	elements? Explain using examp	les.			
4.	Inferring Why could you use clueven though the table is now ba				
5.	Communicating Write a paragrelements are not included in the not you think it is likely that an	alien periodic tab	le. Explain whether or		

More to Explore

Notice that Period 5 is incomplete on the alien periodic table. Create names and symbols for each of the missing elements. Then, compose a series of clues that would allow another student to identify these elements. Make your clues as precise as possible.